# **PROGRAMME AT A GLANCE** Sunday, 15th June 17.00-19.00: Registration 20.00: Welcome reception Monday, 16th June 8.50-9.10: Welcome 9.10-9.50: V. Vivier (Invited lecture) Operando combination of techniques for studying corrosion processes 9.50-10.10: H. Hubin Designing sustainable materials through a combined modelling and operando experimental approach 10.10-10.30: M. Dabiri Havigh Application of operando ORP-EIS for studying anion incorporation during anodizing 10.30-10.50: C. Lamloum (eligible for best presentation award) Study of corrosion of aluminum by coupling acoustic and electrochemical measurements

	10.50-11.20: Coffee break		
	rochemical and al methods	11.20-11.40: <b>B. Sultan</b>	
		Development of a novel coupled analytical approach combining	
		Atomic Emission Spectro electrochemistry (AESEC) and	
		Electrochemical Quartz Crystal Microbalance (EQCM) techniques	
		11.40-12.00: A. Sotniczuk	
		Corrosion behavior of new orthopaedic alloys designed based on	
		Ti-Mo system: real-time AESEC dissolution analysis combined	

with ex-situ TEM characterization 12.00-12.20: F. Di Franco

Synergistic use of electrochemical and photoelectrochemical methods to study the corrosion behaviour of TiAlV alloys

12.20-12.40: J.C. Salvador Fernandes EIS, photoelectrochemical and ellipsometric study of niobium pentoxide deposited on Ti-6Al-4V alloy by DC reactive sputtering

12.40-13.00: K. Fushimi Online ICP-OES measurements for investigating corrosion of Al-10Si alloys in chloride environment

## 13.00-14.30: Lunch

Passivation and localized corrosion of high entropy alloys 14.50-15.10: K. Ogle

14.30-14.50: P. Marcus

The role of alloying elements in the spontaneous passivation of CoCrFeNi multi-principal element alloy (MPEA) family

15.30-15.50: Y. Hou

Investigating the role of Cu and Mo in the corrosion resistance and passivation behavior of equiatomic CoCrFeNi multiprincipal element alloys (MPEA) using in situ detection methods

15.50-16.10: A. Nasiri

Unveiling passivity response of arc directed energy deposited 308 austenitic stainless steel in 0.1 M sulfuric acid solution

16.10-16.30: O. Devos

Corrosion study of wire arc additively manufactured SS 316L -Influence of the mechanical deformation and comparison with conventional 316L

16.30-16.50: M. Leonardi (eligible for best presentation award) Corrosion behavior of pure Ta in a concentrated and hot nitric acid environment

### 16.50-17.30 POSTER SESSION 1 and Coffee break

17.30-17.50: M. Cabrini

Hydrogen solubility and distribution inside spheroidal cast iron 17.50-18.10: N. Macháčková (eligible for best presentation

Efficiency of corrosion-induced hydrogen entry into steel

18.10-18.30: L. Cupertino-Malheiros

New insights into the hydrogen evolution reaction and absorption kinetics using multiple electrochemical techniques

18.30-18.50: C. Pérez

The galvanic couple carbon steel-shape memory steel in media of different aggressiveness

18.50-19.10: M. Motta (eligible for best presentation award)

Chitosan as a corrosion inhibitor for gray cast iron and its role in mitigating the stiction phenomenon in automotive braking systems

20.00: Dinner

## Tuesday, 17th June 8.50-9.30: M.E. Orazem (Invited lecture) On interpretation of impedance spectra 9.30-9.50: S. Martinez Al-aided EIS method for streamlined in situ corrosion rating of bioimplant metallic materials 9.50-10.10: B. Tribollet EIS analysis of oxide layers developed on nickel-based alloy steam generator tubes during exposure to primary water 10.10-10.30: M. Itagaki rocher Impedance analysis on localized corrosion of copper with transmission line model 10.30-10.50: A. Gabryelczyk Electrochemical impedance spectroscopy for characterization of TiO<sub>2</sub> layers: semiconducting and corrosion properties 10.50-11.20: Coffee break 11.20-11.40: Z. Jiryaeisharahi Exploring the diffusion kinetics of water and ions through organic coatings using ORP-EIS 11.40-12.00: **D. Jero** Water uptake and molecular mobility in polyamide coatings: A comparison of 3- and 4-electrode EIS experiments 12.00-12.20: M. Sakairi Investigation into corrosion behavior of metals in artificial sea water at low temperature with electrochemical methods 12.20-12.40: Y. Morozov Electrochemical Frequency Modulation (EFM) technique to quantify the corrosion of aluminium and magnesium alloys 12.40-13.00: L. Mezzomo Correlating electrochemical noise measurements and neutral salt spray tests on primary and recycled anodized aluminumsilicon allovs 13.00-14.30: Lunch 14.30-14.50: H. Terryn The critical role of electrolyte evolution in atmospheric corrosion sensor performance 14.50-15.10: P. Linhardt Electrochemistry in remote places - MICRA-buoy as an instrumented probe for assessing the corrosiveness in natural waters 15.30-15.50: M.L. Zheludkevich Optimising discharge of Mg anode through combination of modelling with electrochemical techniques 15.50-16.10: S. Lamaka Contribution of oxygen reduction reaction to the cathodic processes during magnesium degradation in selected complex environments 16.10-16.30: N. Madelat Real-time monitoring of the influence of current density and temperature on performance of Pb-Ag anodes in zincelectrowinning systems 16.30-16.50: J. Han Revisiting Ni dissolution in sulfuric acid using element-resolved electrochemical impedance spectroscopy 16.50-17.30 POSTER SESSION 2Coffee break 17.30-17.50: E. Martínez-Pañeda Virtual Electro-Chemo-Mechanical experiments to unravel and predict localized corrosion 17.50-18.10: M. Meeusen The coupling of experiments and finite element models for the prediction of corrosion phenomena 18.10-18.30: V. Bongiorno A machine learning approach for the interpretation of

electrochemical data from corroding systems

machine-learning based alloy design

Local and element-resolved electrochemical setup via atomic

emission spectroelectrochemistry coupled with scanning flow

cell: Toward high-quality element-resolved datasets for

Leveraging unsupervised machine learning for advanced

18.30-18.50: N. Acevedo

18.50-19.10: V. Shkirskiy

electrochemical analysis

20.00: Dinner

Wednesday, 18th June 8.50-9.30: A.W. Hassel (Invited lecture) Scanning droplet cell microscopy as a powerful tool in corrosion research 9.30-9.50: E. Kurchavova Electrochemical screening method in ionic liquid microdroplet coupled with goniometer measurement 9.50-10.10: G. Williams New insights into the application of scanning vibrating electrode technology to investigate the localised corrosion of magnesium 10 10-10 30: A C Rastos The influence of vibration and probe movement on SVET measurements 10.30-10.50: L.F.P. Dick In situ current mapping during the crevice corrosion of stainless steel 10.50-11.20: Coffee break 11.20-11.40: M.-G. Olivier Correlations between composition, microstructure and corrosion behavior of AM AlSi7Mg0.6 and AM 7075-Ti and alloys from selective laser melting (SLM): Electrochemical approach to local and global phenomena 11.40-12.00: A. Cristoforetti Exploring the corrosion resistance and microstructural evolution of textured aluminum allovs using TMEMM 12.00-12.20: M. Magnan (eligible for best presentation award) Corrosion evolution and protection provided by the industrial trivalent chromium process of aluminium-copper and aluminium-copper-lithium alloys 12.20-12.40: **H.G.** de Melo Study of the corrosion resistance of 2024-T3 aluminum alloy anodized in TSA with the addition of graphene oxide 12.40-13.00: T. Liao (eligible for best presentation award) Surface modification of anodised aluminium with fibrous silica for enhanced corrosion resistance 13 00-14 30: Lunch 15.00-18.30: Visit to the Murano Glass Museum 20 00: Gala dinner



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# Thursday, 19<sup>th</sup> June

8.30-8.50: I. De Graeve

Electrochemical conversion of additively manufactured aluminium allovs

8.50-9.10: S. Akbarzadeh (eligible for best presentation award)

PEO coatings for additively manufactured AlSi7Mg0.6 alloy: Unveiling the influence of surface morphology

9.10-9.30: K.A. Yasakau

Corrosion degradation mechanisms of PEO-coated aluminum alloys studied by electrochemical impedance spectroscopy

9.30-9.50: O. Gharbi

Exploring cation - anion synergy of environmentally friendly ionic liquids for the corrosion inhibition of Al allovs

9.50-10.10: R. Kaddah (eligible for best presentation award)

Electrochemical characterization of corroded 2024-T3 alloy treated with environmentally friendly carboxylic acids for aerospace applications

10.10-10.30: F.J. Rodriguez-Gómez

Conversion coating with neodymium additions for AS21 alloy for the automotive industry through a deep eutectic solvent (DES)

10.30-11.00: Coffee break

11.00-11.20: A. Kraš (eligible for best presentation award) Response surface methodology for zirconium conversion coating optimisation on cold-rolled steel, Zn, and AA5754 substrates

Mg-Al-LDH pigments dried by lyophilization and loaded with different corrosion inhibitors: tailored corrosion evaluation by electrochemical techniques

11.40-12.00: I.V. Aoki

Lithium and Cerium (III) salts loaded in epoxy coating on galvanized steel to improve corrosion resistance

12.00-12.20: E. Rahimi

Enhanced microstructural, nanomechanical, and corrosion properties of optimized CrN/Cr coatings on carbon steel

12.20-12.40: A. Nasiri (eligible for best presentation award)

Ion implantation: A breakthrough for superior corrosion resistance and sustainable surface solutions

12.40-13.00: D.R. Ceratti

Are halide perovskites corrosive materials? Challenges and insights into the corrosion behaviour of halide perovskite solar panels

13.00-14.30: Closing and lunch

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# **POSTER SESSION 1**

### A. Bastos

Galvanic interactions of copper with oxide films and differential temperature

In-situ monitoring of pitting corrosion in carbon steel for offshore structures using electrochemical techniques

## H. Laieghi (eligible for best poster award)

Synergistic drivers of intergranular corrosion in SLM AlSi10Mg alloys

### G. Lin

Tracking hydrogen in metals: permeation and thermal desorption study using a deuterium tracer

### L. Gritti (eligible for best poster award)

Simplified methodology for estimating diffusion coefficient and hydrogen uptake via data driven optimization approach

### K. Nomura (eligible for best poster award)

Acceleration of atmospheric corrosion rate of iron at low temperature

# J. Yuan

Real-time monitoring of oxygen evolution reaction and elemental dissolution kinetics via AESEC-respirometry: Corrosion mechanisms for bipolar plates in proton exchange membrane electrolysis

## H.G. de Melo

Impact of microstructure on the corrosion resistance of a novel aluminium crossover alloy

## D. Asperti (eligible for best poster award)

Effect of heat treatment on corrosion resistance of AlSi9Cu3 alloy produced by laser powder bed fusion

## A. Simons

Innovative methodology for the use of electrochemistry in corrosion research in heat pumps using a H<sub>2</sub>O-NH<sub>3</sub> coolant

Study of nitric acid reduction mechanisms on a silicon-enriched austenitic stainless steel under heat transfer conditions: a technological challenge

## M. Mohammedture (eligible for best poster award)

The effect of selective dissolution on electrochemical impedance spectra of 3D-printed aluminum alloys subjected to chlorideinduced corrosion

## T. Cordoba

Corrosion resistance of Ferritic Stainless Steels pickled by laser

Electrochemical behaviour of 42CrMo4 QT steel in commercial and pure methanol

### W. A. Khan (eligible for best poster award)

Electrochemical methods and microstructural characterization to investigate the corrosion resistance properties of 316L samples printed through Binder Jetting and sintered using different furnaces

## H.G. de Melo

Comparing the corrosion resistance in the as-printed condition of 17-4 PH stainless steel obtained from two additive manufacturing technologies

## C. Hejjaj

Investigating the impact of partitioning behavior of a C16 surfactant on the corrosion protection of steel in a CO<sub>2</sub>-saturated environment



# **POSTER SESSION 2**

## M. Macht (eligible for best poster award)

Electrochemical behavior of steel in alternative cement mixtures: insights into corrosion and passivation dynamics

Four-electrode impedance measurement for the corrosion monitoring of steel in concrete

## A. Makogon (eligible for best poster award)

Automated hydrogen permeation analysis using machine learning-enhanced optical microscopy

Producing plasma electrolytic oxidation (PEO) corrosion resistant coatings on aluminium 2024 texturized with a riblet-like surface for aeronautical applications

### N. Kovač

Corrosion protection of AZ31 magnesium alloy using sol-gel coating for biomedical applications

### A. Olesiński

Evaluation of corrosion resistance and mechanical properties of particle-doped coatings produced in plasma electrolytic oxidation under soft-sparking regime

## O. Bannour (eligible for best poster award)

Electrosynthesis of Zn-M (M=Cu, Sn, Mg, Al) protective coatings for enhanced steel corrosion resistance

Development of 3D conductive cement-based anodes for impressed current cathodic protection of reinforced concrete

## H.G. de Melo

Corrosion performance of AA7075-T6 alloy anodizing in a sulphuric-tartaric acid bath and polymer sealing

The effect of cerium organophosphate corrosion inhibitor on water absorption in epoxy-based coatings assessed by electrochemical impedance

## E. De Ketelaere

Comprehensive electrochemical characterization of silicate-based corrosion inhibition

### S. Al Awadh (eligible for best poster award)

Detection of localized corrosion in steel samples immersed in soil using electrochemical noise and shot noise analysis

## E. Tubaro (eligible for best poster award)

Electrochemical evaluation of formulation parameters in epoxy primers for fouling-release systems resistant to cathodic disbonding on bronze alloys

### D. Veys-Renaux

Sealing performances on anodic layers: comparative evaluation by electrochemical impedance spectroscopy and standard

### D. Parasińska

Amplitude-modulated excitation in nonlinear electrochemical impedance spectroscopy: comparative insights from galvanostatic and potentiostatic measurements for corrosion monitoring

Evaluation of corrosion behaviour of Zn-Al alloys in simulated concrete environments using electrochemical impedance spectroscopy

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